UNCLASSIFIED

AD 296 269

Reproduced by the

ARMED SERVICES TECHNICAL INFORMATION AGENCY
ARLINGTON HALL STATION
ARLINGTON 12, VIRGINIA



UNCLASSIFIED

NOTICE: When government or other drawings, specifications or other data are used for any purpose other than in connection with a definitely related government procurement operation, the U. S. Government thereby incurs no responsibility, nor any obligation whatsoever; and the fact that the Government may have formulated, furnished, or in any way supplied the said drawings, specifications, or other data is not to be regarded by implication or otherwise as in any manner licensing the holder or any other person or corporation, or conveying any rights or permission to manufacture, use or sell any patented invention that may in any way be related thereto.

CROSTRUCTURE STUDIES OF POLYCRYSTALLINE REFRACTORY OXIDES QUARTERLY PROGRESS REPORT NO. 2

Prepared by

T. Vasilos
R.M. Spriggs
J.B. Mitchell

RAD-SR-63-29
Prepared under U.S. Navy, Bureau of Weapons
Contract Now 62-0648c

14 February 1963

FEB 1.9 1983

1 1 1

SINCE THIS IS A PROGRESS REPORT, THE INFORMATION CONTAINED HEREIN IS TENTATIVE AND SUBJECT TO CHANGES, CORRECTIONS AND MODIFICATIONS. QUALIFIED REQUESTORS MAY OBTAIN COPIES OF THIS REPORT DIRECT FROM ASTIA.

RESEARCH AND ADVANCED DEVELOPMENT DIVISION AVCO CORPORATION
Wilmington, Massachusetts

MICROSTRUCTURE STUDIES OF POLYCRYSTALLINE REFRACTORY OXIDES QUARTERLY PROGRESS REPORT NO. 2

Prepared by

T. VasilosR. M. SpriggsJ. B. Mitchell

RAD-SR-63-29
Prepared under U.S. Navy, Bureau of Weapons
Contract Now 62-0648c

14 February 1963

APPROVAL

Charles Spencer, Manager Materials Department

SINCE THIS IS A PROGRESS REPORT, THE INFORMATION CONTAINED HEREIN IS TENTATIVE AND SUBJECT TO CHANGES, CORRECTIONS AND MODIFICATIONS. QUALIFIED REQUESTORS MAY OBTAIN COPIES OF THIS REPORT DIRECT FROM ASTIA.

RESEARCH AND ADVANCED DEVELOPMENT DIVISION AVCO CORPORATION
Wilmington, Massachusetts

ABSTRACT

Transverse bend strength and elastic modulus determinations made of fine-grained (1 to 2 micron), dense, pure Al₂O₃ and MgO specimens as a function of temperature (up to 1350 °C) revealed values higher than those obtained by other investigators for these oxides in more porous form and of larger grain sizes. Studies of the grain growth kinetics of alumina and magnesia were conducted in order to obtain information to enable preparation of specimens with desired larger grain sizes for subsequent thermomechanical testing. Based on observations of the fracture modes exhibited by alumina and magnesia as a function of temperature, it was suggested that anelastic deformation or possible plastic flow in the so-called "equicohesive" temperature range may influence the mode of fracture. By analogy with findings in metals systems, it was hypothesized that the apparent transition from transgranular to intergranular fracture might occur below some critical stress rather than above some critical temperature.

CONTENTS

I.	Introduction	1
II.	Grain Growth Studies	2
III.	Mechanical Properties Determinations	6
ıv.	Observations on Fracture Modes	11
v.	Future Work	16
VI.	References	17

ILLUSTRATIONS

Figure	1	Log Grain Size versus Log Time for Grain Growth of MgO and Al ₂ O ₃	3
	2	Typical Micrographs of Al ₂ O ₃ Specimens Heat-Treated at 1500°C to Cause Grain Growth to Occur	4
	3	Typical Micrographs of MgO Specimens Heat-Treated to Cause Grain Growth to Occur	5
	4	Load versus Deflection for Transverse Bending at Various Test Temperatures	8
	5	Modulus of Elasticity versus Temperature	9
	6	Modulus of Rupture versus Temperature	10
	7	Typical Fracture Surfaces Exhibited by Specimens of Al ₂ O ₃ and MgO Broken in Transverse Bending at Various Test Temperatures	13
	8	Electron Fractographs of Al ₂ O ₃ Specimens Broken in Transverse Bending at Various Temperatures	14
	9	Electron Fractographs of MgO Specimens Broken in Transverse Bending at Various Temperatures	15

I. INTRODUCTION

This is the second quarterly progress report on microstructure studies of polycrystalline refractory oxides. The overall objective of this project is to determine the effects of microstructure on the elevated-temperature, mechanical-strength properties of selected ceramic-oxide refractory materials. Primary emphasis is currently being placed on the effect of grain size on the elevated-temperature transverse bend strength and elastic modulus of pure, single-phase alumina and magnesia.

During the period covered by this report, transverse bend strength and elastic modulus determinations were made on fine-grained (1- to 2-micron) dense, pure Al₂O₃ and MgO as a function of temperature. In addition, preliminary studies were conducted to determine the kinetics of grain growth of these materials in order to obtain the desired larger grain sizes for this investigation. Finally, some observations were made of the modes of fracture, exhibited by the fine-grained alumina and magnesia as a function of temperature. The influence of anelastic deformation or possible plastic flow on the mode of fracture in the "equicohesive" temperature range has been discussed.

In this report, the bend strength and elastic modulus values obtained for fine-grained alumina and magnesia are compared with earlier data on these materials reported by Schwartz¹ and Coble and Kingery². It is recognized that several other investigators (see, for example, references 3 through 6) have also studied the room and elevated temperature bend strength and elastic modulus behavior of these materials with varying microstructures, purities, testing conditions, etc. However, data were not available for simultaneous measurements of strength and modulus on the same, pure, highly dense specimens as a function of grain size and temperature. Thus, the data of Schwartz and Coble and Kingery proved to be most suitable for comparison with the present results.

Possible effects due to factors such as surface conditions continue to be under close surveillance during this study. For example, surface-roughness measurements made on various Al₂O₃ and MgO specimens before and after elevated-temperature testing have revealed no measurable changes in surface condition during testing.

Future plans call for completing the bend strength and elastic modulus determinations as a function of temperature of alumina and magnesia specimens which have received thermal treatments to cause varying amounts of grain growth to occur.

II. GRAIN GROWTH STUDIES

In order to obtain the desired grain sizes for this investigation, several preliminary studies were made to determine the kinetics of grain growth of the alumina and magnesia materials. Figure 1 plots log grain size versus log time for the grain growth of Al_2O_3 and MgO at various temperatures. The grain-size determinations were made with an optical microscope using a filar eyepiece. In the early stages of grain growth, the grain sizes were at the limit of resolution of the microscope and considerable error resulted from an inability to resolve small individual grains. Accurate determinations of the smaller grain sizes are now being made from electron-microscope replicas of the specimens. The curves shown in figure 1 are still useful in determining the time and temperature to obtain a larger grain size. Specimens are now being heat-treated to obtain the larger grain sizes for testing. Examples of the microstructures obtained during grain growth of Al_2O_3 and MgO are shown in figures 2 and 3, respectively.

The rather slow grain growth rate obtained in Al₂O₃ has led to efforts to enhance the growth by heating in a hydrogen atmosphere with a subsequent treatment in air. These experiments are now being carried out.

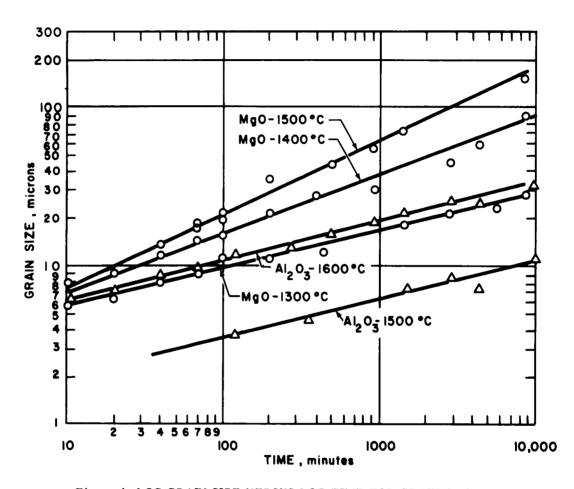


Figure 1 LOG GRAIN SIZE VERSUS LOG TIME FOR GRAIN GROWTH OF MgO AND A1 $_2\text{O}_3$ $_{63-1548}$

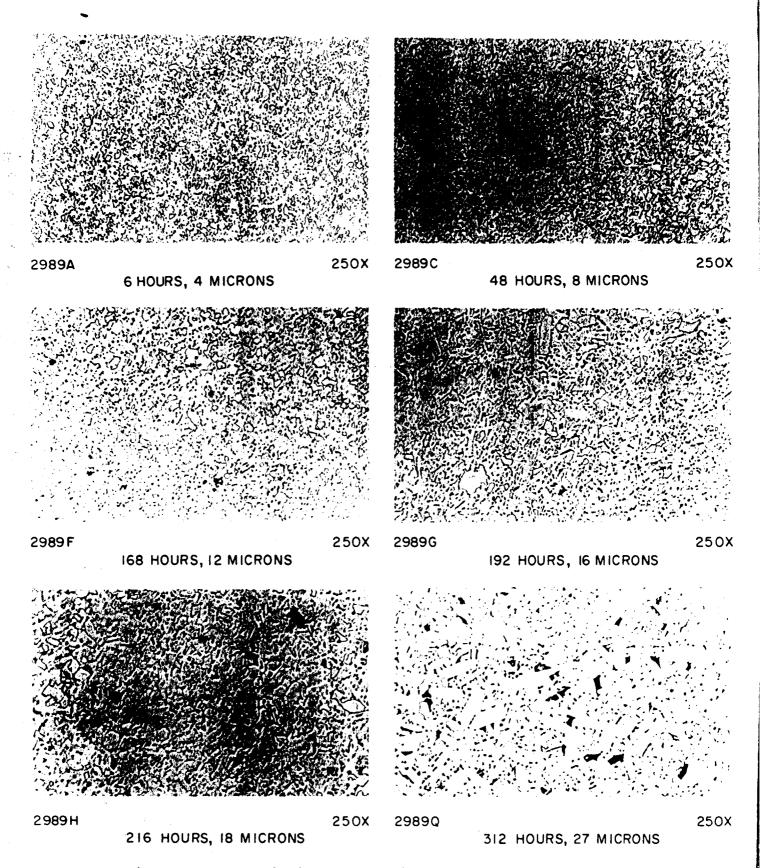


Figure 2 TYPICAL MICROGRAPHS OF A12O3 SPECIMENS HEAT-TREATED AT 1500°C TO CAUSE GRAIN GROWTH TO OCCUR
63-1553

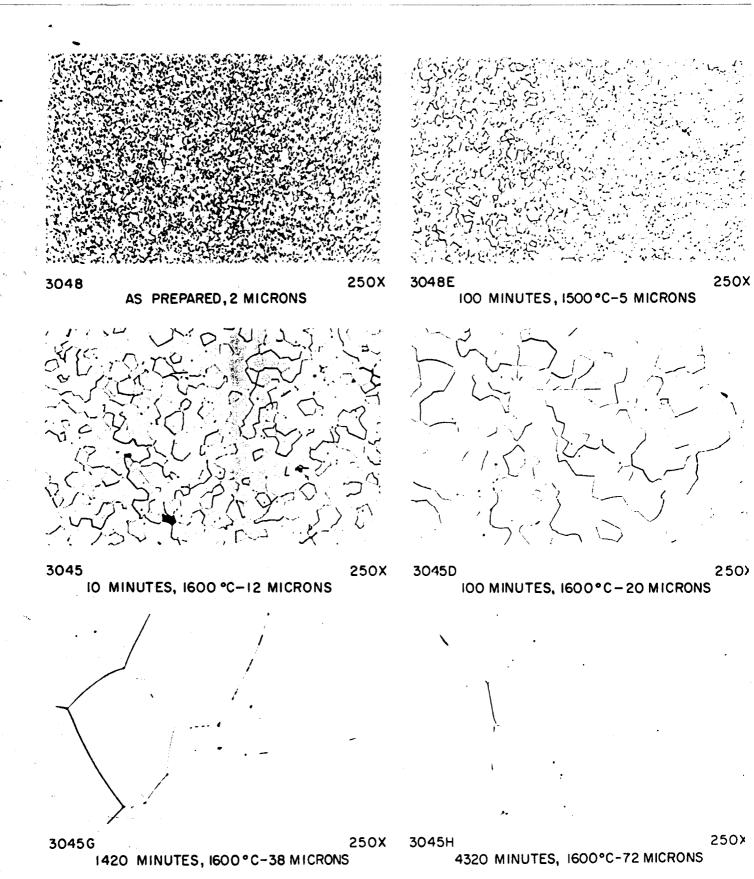


Figure 3 TYPICAL MICROGRAPHS OF MgO SPECIMENS HEAT-TREATED TO CAUSE GRAIN GROWTH TO OCCUR
63-1551

III. MECHANICAL PROPERTIES DETERMINATIONS

The initial tests of elastic modulus in transverse bending yielded erratic results. It was found that the correction for apparatus deflective at the test temperature changed in an unpredictable manner with each application of the load, due to thermal and stress effects on the apparatus parts. This problem was eliminated by modifying the apparatus so that the deflection-sensing instrument (linear variable differential transformer) measured only specimen deflection with respect to the lower knife edges, eliminating the need of correcting for apparatus deflection.

Figure 4 shows typical load-deflection curves obtained for transverse bending of fine-grained (1- to 2-micron) Al₂O₃ and MgO at various test temperatures. Elastic modulus and modulus of rupture values were calculated with the conventional beam-deflection formulas in four-point loading:

$$E_{t,b.} = \frac{P}{v} \left[\frac{a}{4 \text{ bd}^3} (3l^2 - 4a^2) \right]$$

$$\sigma_{t,b.} = \frac{3 Pa}{4 \text{ bd}^2}$$

where

P = load, pounds

l = span, inches

b = specimen width, inches

d = specimen depth, inches

y = specimen deflection, inches

a = distance from point of load to point of reaction, inches.

The ratio P/y was determined from the linear portion of the load-deflection curve. The elastic modulus and modulus of rupture of fine-grained Al₂O₃ and MgO are shown as a function of test temperature in figures 5 and 6, respectively. Also plotted for comparison are the data of Schwartz¹ and Coble and Kingery². The curves of elastic modulus are in general agreement with the trend of the data presented by the above investigators. Both the alumina and magnesia exhibited higher elastic moduli and showed stronger temperature dependence between 30 and 400°C and between 700 and 1000°C than the materials used by Schwartz and Coble. It is not yet clear whether this effect was due principally to a decreased grain size or a decrease in porosity. Coble's

alumina had a grain size of 25 microns and a minimum of 10 percent porosity. The alumina used by Schwartz had 4 percent porosity and the magnesia 11 percent porosity. The grain size of Schwartz's materials was not known.

While several other investigators have studied microstructure effects on the elevated-temperature mechanical properties of refractory oxides, 3-6 the data of Schwartz and Coble and Kingery proved to be most suitable for comparison with the present results.

Comparison of the transverse bend strengths of Al₂O₃ and MgO with those of Schwartz showed a significant increase in strength which was probably due to a combination of higher density and smaller grain size in the present materials. However, an exact quantitative comparison of the strength values is not justified since the specimen cross section and gage lengths used by Schwartz differed for those used in the present investigation.

The large degree of scatter in the reported data was principally due to differences in fabrication conditions, which were initially varied to achieve optimum density, grain size, and strength in the specimens. Specimens fabricated under the same conditions of temperature, pressure, and time exhibited relatively little scatter in mechanical properties. Specimens fabricated under slightly different conditions, e.g., relatively higher hot-pressing temperature but for shorter time, although having the same apparent density and grain size, exhibited significant differences in transverse bend strength and elastic modulus. It is not, at present, understood how these variations in fabrication conditions affect the mechanical properties. Subsequent specimens for this investigation, approximately 400 of alumina and 300 of magnesia, were fabricated under identical conditions. Testing of specimens having a grain size of approximately 50 microns is now being carried out.

Testing of surface conditions of the specimens before and after high-temperature mechanical testing was accomplished by running surface-roughness tests on a Talysurf machine. The average deviation from a reference centerline (centerline average, CLA) was taken as the measure of surface condition. No change was found to occur in the surface conditions of the specimens during testing. The average surface roughness was 14 to 22 μ inches.

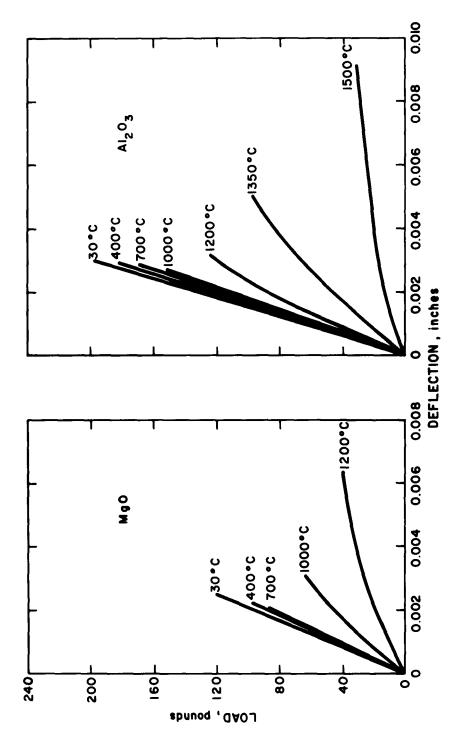


Figure 4 LOAD VERSUS DEFLECTION FOR TRANSVERSE BENDING AT VARIOUS TEST TEMPERATURES 63-1547

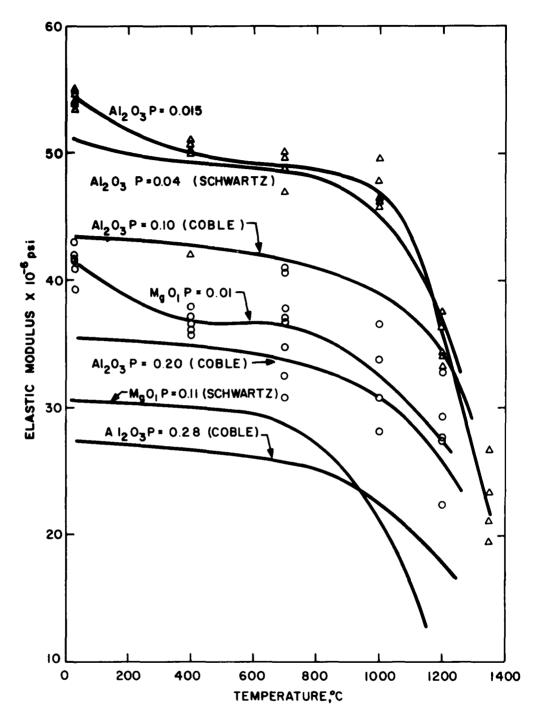


Figure 5. MODULUS OF ELASTICHY VERSUS TEMPERATURE $63\cdot1546$

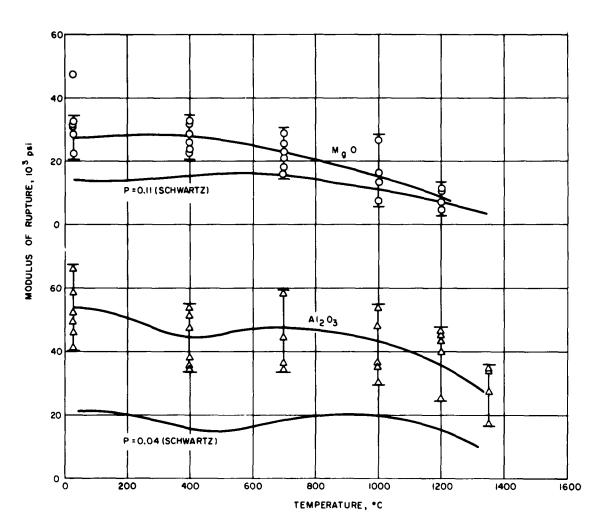


Figure 6 MODULUS OF RUPTURE VERSUS TEMPERATURE $6.3 \cdot 1.549$

IV. OBSERVATIONS ON FRACTURE MODES

Typical fracture surfaces exhibited by Al₂O₃ and MgO specimens broken at various test temperatures are shown in figure 7. The alumina specimens exhibited very rough and irregular fracture surfaces up to 1200 °C and magnesia up to 700 °C. Specimens fractured at room temperature frequently broke into three or four pieces in the local region of the fracture. At higher temperatures, the fractures became smooth and regular in appearance as though sliced normal to the tensile axis.

Figures 8 and 9 are electron fractographs of the fractured surfaces of several A12O3 and MgO specimens. These fractographs revealed a combination of intergranular and transgranular fracture in both A12O3 and MgO. The principal mode of fracture in A12O3 at test temperatures up to 1000°C and in MgO up to 700°C appeared to be transgranular. Intergranular fracture was also observed in the microstructure in this temperature range. Fractographs of specimens tested at higher temperatures revealed fracture to be predominantly intergranular. Fracture in A12O3 specimens tested at 1350°C and in MgO specimens tested at 1200°C appeared to be entirely intergranular.

These observations indicated that the change in macroscopic fracture characteristics exhibited in Al₂O₃ and MgO over the so-called "equicohesive" temperature range (see, for example, reference 7) results from an increase in the ratio of intergranular to transgranular fracture with increasing test temperature.

It is of interest to note that it is in these temperature ranges that the specimens begin to exhibit yielding and a nonlinear load-deflection behavior prior to fracture. It is suggested that this anelastic deformation or possible plastic flow may influence the mode of fracture of these materials. Strain damage in and about the grain-boundary region might lead to intergranular fracture. Chang and Grant⁸ found that stress concentrations at the junction of grains where grain-boundary sliding had occurred often resulted in the formation of cracks. They found that cracks spread from one triple point to another along a more or less direct path in the plane of the boundary. McLean, 9-12 in a series of unique experiments showed that creep deformation in polycrystalline metal aggregates occurred by means of migration of dislocations (resulting in slip and subboundary formation) and by means of grain-boundary sliding. Under a given stress, the ratio of the fraction of creep strain arising from grain-

boundary shearing to total creep strain $\frac{E_{g.b.}}{E_{t}}$ remained essentially constant.

This ratio was found to increase as the stress decreased. Using McLean's technique on creep in polycrystalline aluminum, Fazan, Sherby, and $Dorn^{13}$

confirmed McLean's observations and showed that the ratio $\frac{E_{g.b.}}{E_{t}}$ for a given

stress was independent of temperature. As suggested by Dorn, ¹⁴ these results indicated that grain-boundary shearing might be attributed to localized crystallographic mechanisms of deformation in the region of the grain boundary rather than a process such as viscous shearing. Since the strain arising from grain-boundary shearing is a function of the total strain for a given stress, independent of temperature, the localized strain damage in and around the grain-boundary region such as might lead to intergranular fracture should also depend on the total strain independent of temperature. McLean's data which

showed that the ratio $\frac{E_{g.b.}}{E_{t}}$ increases with decreasing stress suggests that

the strain damage becomes more and more concentrated in the vicinity of the grain boundary as the stress is decreased.

While similar observations of structural changes attending deformation in fully dense polycrystalline refractory oxides have not been made, it is reasonable to assume that the same relationships might be evident in these materials. Such observations in these materials would indicate that the apparent transition from transgranular to intergranular fracture might occur below some critical stress rather than above some critical temperature.

A complimentary study of phenomenological and structural changes attending creep deformation in fully dense MgO and Al₂O₃ would elucidate possible relationships between deformation, microstructure, and mechanical properties of refractory oxides.

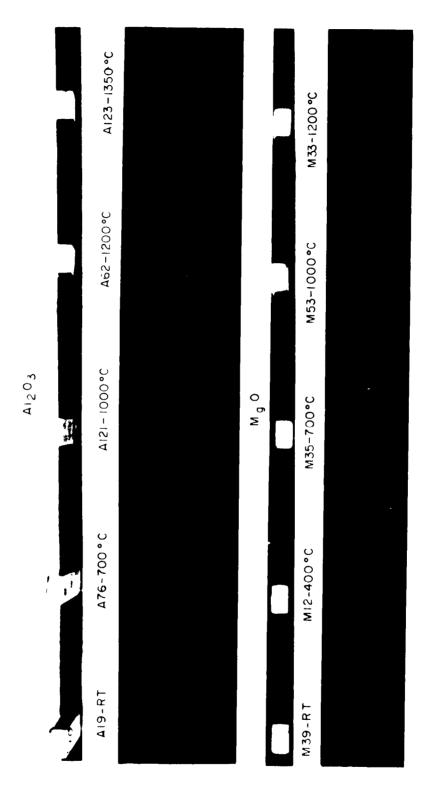


Figure 7 TYPICAL FRACTURE SURFACES EXHIBITED BY SPECIMENS OF A1₂O₃ AND MgO BROKEN IN TRANSVERSE BENDING AT VARIOUS TEST TEMPERATURES 63-1550

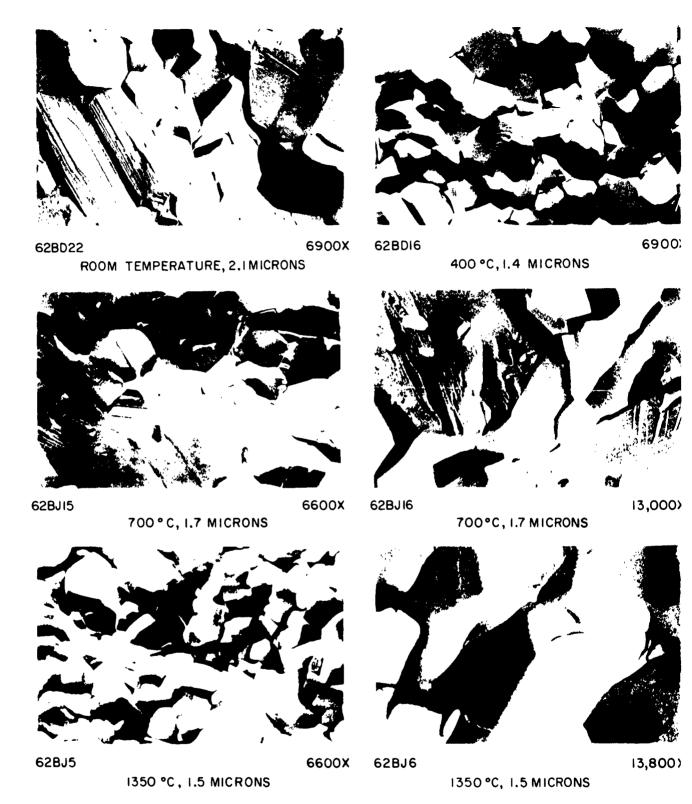
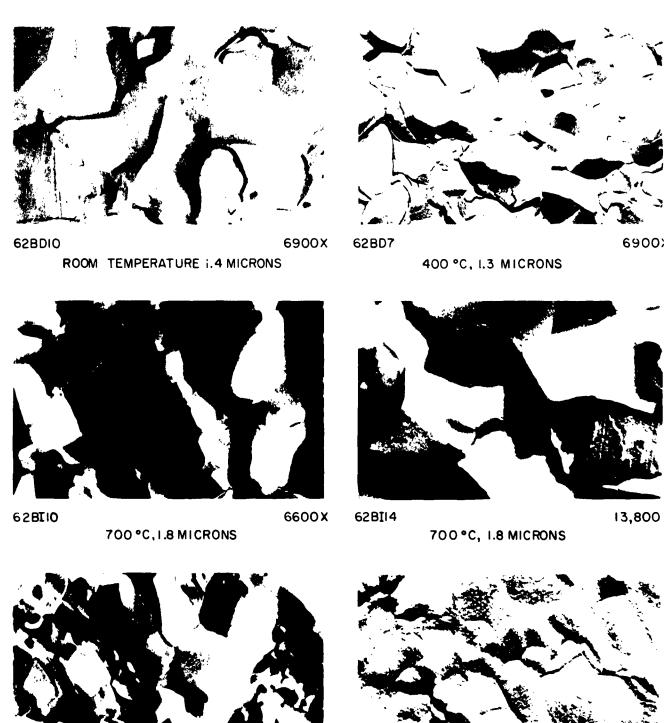


Figure 8 ELECTRON FRACTOGRAPHS OF $A1_2O_3$ SPECIMENS BROKEN IN TRANSVERSE BENDING AT VARIOUS TEMPERATURES 63-1552



62**B**IO 13,800> 6600 X 62BI5

Figure 9 ELECTRON FRACTOGRAPHS OF MgO SPECIMENS BROKEN IN TRANSVERSE BENDING AT VARIOUS TEMPERATURES 63-1554

1200 °C, 2.0 MICRONS

1200°C, 2.0 MICRONS

V. FUTURE WORK

The necessary specimens of fine-grained, dense, pure Al₂O₃ and MgO to cover the desired larger grain-size ranges and testing temperatures have been fabricated and machined. Subsequent thermal treatments are being employed to achieve the larger grain sizes up to 300 to 500 microns. During the remainder of the project, it is planned to complete the strength and modulus determinations of these larger grain-size specimens as a function of temperature. Testing of specimens having a grain size of approximately 50 microns is now in progress.

VI. REFERENCES

- Schwartz, B., Thermal Stress Failure of Pure Refractory Oxides,
 J. Am. Ceram. Soc., 35 (12) 325-33 (1952).
- 2. Coble, R. L., and W. D. Kingery, Effect of Porosity on Physical Properties of Sintered Alumina, J. Am. Ceram. Soc., 39 (11) (1956).
- Crandall, W. B., D. H. Chung, and T. J. Gray, The Mechanical Properties of Ultrafine Hot Pressed Alumina, Proceedings of Conference on Mechanical Properties of Engineering Ceramics, Interscience Publishers, New York (1961), p. 345-379.
- 4. Harrison, W. B., Fracture Behavior of Dense Polycrystalline MgO, Presented at 63rd Annual Meeting of the American Ceramic Society, 24 April 1961).
- 5. Weil, N. A., ed., Studies of the Brittle Behavior of Ceramic Materials, ASD Technical Documentary Report ASD-TR-61-628 (April 1962): (Note particularly Task 9, by R. J. Charles.)
- 6. Johnson, J. E., A. K. Smalley, and W. H. Duckworth, Investigation of Sinterable Powders and Properties of Beryllia Ceramics, WADD Technical Report 60-108 (April 1960).
- 7. Tunison and Burdick, Fracture Analysis of Alumina using the Electron Microscope, Final Report on ONR, Alfred University, Alfred, New York (1962).
- 8. Chang, H. C., and M. J. Grand, Grain Boundary Migration and Intercrystalline Failure Under Creep Conditions, Trans. Am. Inst. Mining Met. Engrs., 197, 304-12 (1953).
- 9. McLean, D., Creep Processes in Coarse Grained Aluminum, J. Inst. Metals, 80, 507-519 (1951).
- 10. McLean, D., Crystal Slip in Aluminum During Creep J. Inst. Metals, 81, 133-144 (1952).
- 11. McLean, D., Crystal Fragmentation in Aluminum During Creep, J. Inst. Metals, 81, 287-292 (1952).

- 12. McLean, D., Grain Boundary Slip During Creep of Aluminum, J. Inst. Metals, 81, 293-300 (1952).
- 13. Fazan, B., O. D. Sherby, and J. E. Dorn, Data recorded in J. Mech. and Phys. Solids 3, 85-116 (1954).
- 14. Dorn, J. E., Some Fundamental Experiments on High Temperature Creep, J. Mech. and Phys. Solids, 3, 85-116 (1954).

DISTRIBUTION

Addressee	No. of Copie	
Bureau of Naval Weapons Department of the Navy Washington 25, D.C.		
Attn: RRMA-34	1	
Attn: RRMA-221	1	
Attn: RMGS-811	1	
Attn: DLI-3	2	
Office of Naval Research		
Department of the Navy		
Washington 25, D. C.		
Attn: Code 423	1	
Naval Research Laboratory		
Washington 25, D. C.		
Attn: Code 6212	1	
Armed Services Technical Information Agency		
Arlington Hall Station		
Arlington, Virginia	10	
National Aeronautics and Space Administration		
1512 H Street N. W.	_	
Washington 25, D. C.	1	
Lewis Research Center, NASA		
Materials and Structures Division		
21000 Brookpark Road		
Cleveland 35, Ohio	1	
Aeronautical Systems Division		
Ceramics and Graphite Branch		
Wright-Patterson Air Force Base, Ohio	2	
Atomic Energy Commission		
Technical Information Service		
P. O. Box 62	_	
Oak Ridge, Tennessee	1	
Army Ballistic Missile Agency		
Huntsville, Alabama		
Attn: Dr. C. H. Reissig	1	

DISTRIBUTION (Cont'd)

Addressee	No.	of Copies
National Bureau of Standards		
Ceramics Section		
Washington 25, D. C.		1
Hdqs., Quartermaster R&E Command		
QM Research and Engineering Center		
Mechanical Engineering Division		
Natick, Massachusetts		1
Armour Research Foundation of Illinois Institute of Technology	Ţ	
10 West 35th Street		
Chicago 16, Illinois		_
Attn: Metals and Ceramics Division		1
Attn: Mechanics Research Division		1
Materials Research Corporation		
47 Buena Vista Avenue		
Yonkers, New York		
Attn: Dr. G. T. Murray		1
Minneapolis-Honeywell Regulator Company		
Research Center		
500 Washington Avenue S.		
Hopkins, Minnesota		1
Massachusetts Institute of Technology		
Cambridge 39, Massachusetts		
Attn: Dr. W. D. Kingery		1
Attn: Dr. Egon Orowan		1
Atomics International Division		
North American Aviation, Inc.		
P.O. Box 309		
Canoga Park, California		1
Battelle Memorial Institute		
Ceramics Division		
505 King Avenue		
Columbus 1, Ohio		1

DISTRIBUTION (Cont'd)

Addressee	No. of Copies
Boeing Airplane Company Aerospace Division	
P.O. Box 3707	
Seattle 24, Washington	•
Attn: Mr. W. M. Sterry	1
School of Ceramics	
Rutgers, The State University	_
New Brunswick, New Jersey	1
Vitro Laboratories	
West Orange Laboratory	
200 Pleasant Valley Way	
West Orange, New Jersey	_
Attn: Dr. S. Grand	1
Gladding, McBean and Company	
2901 Los Feliz Boulevard	
Los Angeles 39, California	_
Attn: Dr. G. M. Butler	1
National Beryllia Corporation	
First and Haskell Avenues	
Haskell, New Jersey	1
Stauffer Chemical Company	
Stauffer Metals Division	
1201 S. 47th Street	
Richmond 4, California	
Attn: J. M. Fitzpatrick	1
State University of New York	
College of Ceramics at Alfred University	
Alfred, New York	1
Pennsylvania State University	
College of Mineral Industries	
University Park, Pennsylvania	
Attn: Prof. F. A. Hummel	1

DISTRIBUTION (Concl'd)

Addressee	No. of Copies
Carborundum Company Research and Development Division Niagara Falls, New York Attn: Dr. W. A. Lambertson	1
Georgia Institute of Technology Engineering Experimental Station Atlanta, Georgia Attn: J. D. Walton	1
Corning Glass Works Corning, New York	1
Coors Porcelain Company Research Department Golden, Colorado	1
University of Utah Department of Ceramics Engineering Salt Lake City, Utah Attn: Dr. I. B. Cutler	1
General Electric Research Laboratories Metallurgy and Ceramics Research Department Schenectady, New York Attn: Dr. J. E. Burke	1
Narmco Research and Development 3540 Aero Court San Diego 23, California Attn: Mr. Roger Long	1
Signal Corps Engineering Labs Fort Monmouth, New Jersey Attn: Mr. S. DiVita	1
Central Files Document Control Research Library	1 5 12